Supplementary materials

Novel polyurethane network/organoclay nanocomposites: Microstructure and physicochemical properties

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Figure 1s. Photographs of the prepared PUN-NC materials.

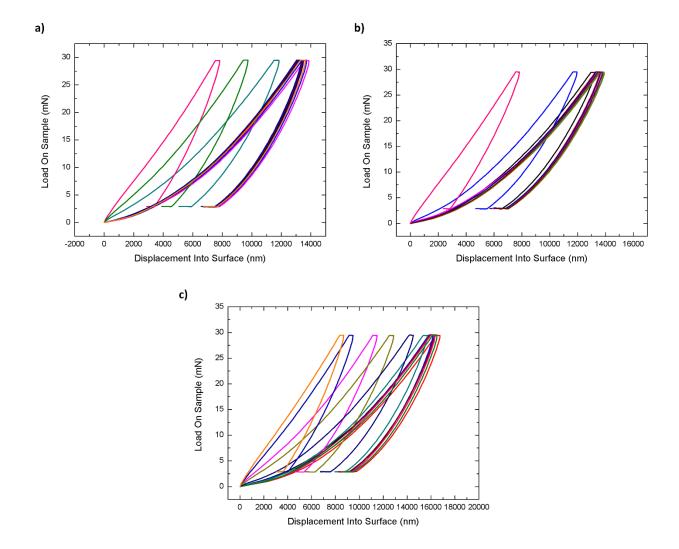


Figure 2s. Curves obtained from nanoindentation measurements for PUN-NC films with 60 wt% of soft segment and different organoclay contents: a) 0.5 wt%; b) 1 wt% and c) 2 wt%.

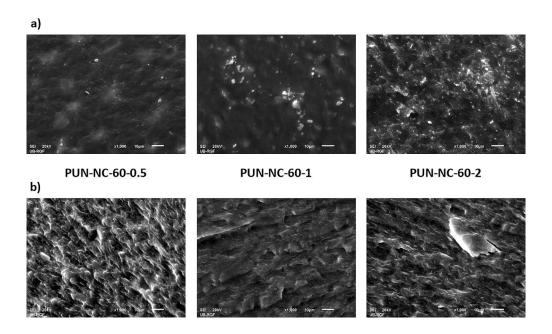


Figure 3s. SEM micrographs of (a) surface and (b) fractured surface PUN-NCs based on different organoclay contents (0.5, 1 and 2 wt%) and with 60 wt% of soft segment at magnification x1000.

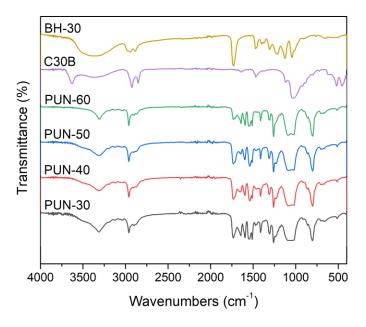


Figure 4s. FTIR spectra of Cloisite 30B, BH-30 and pure PU networks (PUN) with different soft segment contents.

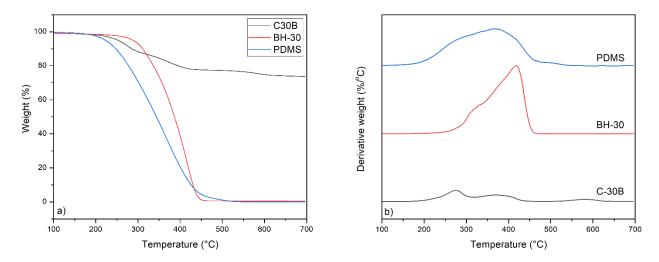


Figure 5s. (a) TGA and (b) DTG thermograms of PDMS, BH-30 and Cloisite 30B determined at a heating rate of 5 °C/min, in nitrogen atmosphere.

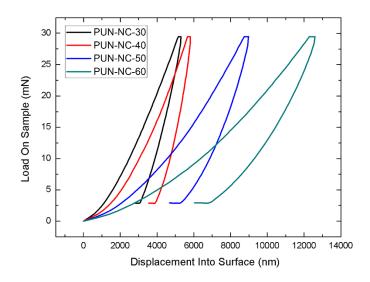


Figure 6s. Typical indentation load-displacement curves for PUN-NC films with different soft segment contents.

Table 1sNanoindentation measurements for PUN-NC with 60 wt% of soft segment and different organoclay contents (0.5, 1 and 2 wt%)

Material	Young	Hardness (MPa)	Plasticity
	modulus		(ductility
	(MPa)		index)
PUN-NC-60-0.5	119 ± 39	14 ± 9	8.5
PUN-NC-60-1	103 ± 39	15 ± 10	6.9
PUN-NC-60-2	94 ± 37	12 ± 9	7.8

Table 2sMass loss values after in *vitro* biodegradation process of the prepared nanocomposites (PUNNC) and pure PU networks (PUN).

Material	Mass loss after three months (wt%)	Mass loss after six months (wt%)
PUN-NC-30	7.01	7.53
PUN-NC-40	7.42	7.80
PUN-NC-50	8.65	14.78
PUN-NC-60	3.88	14.48
PUN-30	1.44	1.44
PUN-40	5.91	18.35
PUN-50	Broke apart	Broke apart
PUN-60	6.90	18.66

Physical parameters of the surface free energy of a solid (γ_S) were found based on the Owens-Wendt method. The method assumes that the surface free energy (γ_S) can be presented as a sum of two components according to equation 1s [48]:

$$\gamma_{\rm S} = \gamma_{\rm S}^d + \gamma_{\rm S}^p \tag{Eq. 1s}$$

where:

 γ_S^d - surface energy connected with dispersion interactions,

 $\gamma_{\rm S}^{p}$ - surface energy connected with polar interactions.

The surface free energy for solids (S) and liquids (L) interacting with those solids should fulfill the Owens-Wendt equation (2s):

$$\gamma_L \cdot \frac{1 + \cos\theta}{2} = \sqrt{\gamma_S^d \cdot \gamma_L^d} + \sqrt{\gamma_S^p \cdot \gamma_L^p}$$
 (Eq. 2s)

where θ is the experimentally found contact angle between a liquid drop and a solid surface under investigation; γ_L is the surface tension of the test liquid. Therefore, the wetting angles θ were first measured for the surfaces of PUN-NC films with the use of model liquids (water and diiodomethane) with the known parameters: $\gamma_L = 72.8$; $\gamma_L^d = 21.8$; $\gamma_L^p = 51.0$ mJ/m² (for water) and $\gamma_L = 50.8$; $\gamma_L^d = 48.5$; $\gamma_L^p = 2.3$ mJ/m² (diiodomethane).

Then, Eq. (2s) was used to calculate the values γ_S^p and γ_S^d for the prepared PUN-NC films. The γ_S values were calculated from Eq. (1s).