FREE RADICAL CARBOCYCLIC RING RECONSTRUCTION

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<u>Abstract</u>: Alkenyl radical generated by *p*-fragmentation of tertiary cyclohexyloxy radical with carbocyclic ring opening, possessing a suitably located olefinic double bond, undergoes to the intramolecular 5-exo-trigonal cyclization and a new carbocyclic ring was formed.

Among a variety of reactions β -fragmentation is one of the preferential mode of stabilisation of tertiary alkoxy radicals¹. Homopolar nature of β -cleavage reaction of alkoxy radicals is well-established and the carbonyl containing fragment and products derived from the alkyl radical fragment were produced¹⁻⁶. The rate of β -cleavage of alkoxy radicals is independent of the radical precursors but is mainly dependent of the stability of the initialy formed carbon radicals^{2,3}. Thereby, intramolecular cyclization of 5-hexenyl radicals to the cyclopentylmethyl radicals are well known reaction for the construction of carbocyclic^{7,8} and heterocyclic rings⁹.

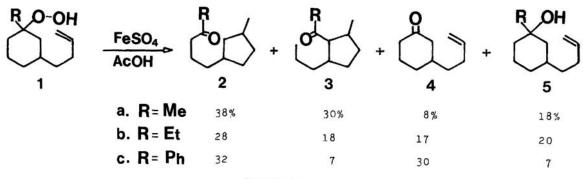
However, homopolar scission of C-C bond in the alkcxy radical intermediates, followed by radical recombination reaction with a new C-C bond formation in the same molecule have not been investigated together as a sequence of reactions. This radical recombination reactions could involve one ring opening and the other ring closure with a considerable change of the carbon skeleton (Scheme 1.).



Scheme 1.

We wish to report a new approach to carbocyclic ring reconstruction involving a tertiary alkoxy radical fragmentation and intramolecular addition of arising alkenyl radical onto the suitably located olefinic double bond.

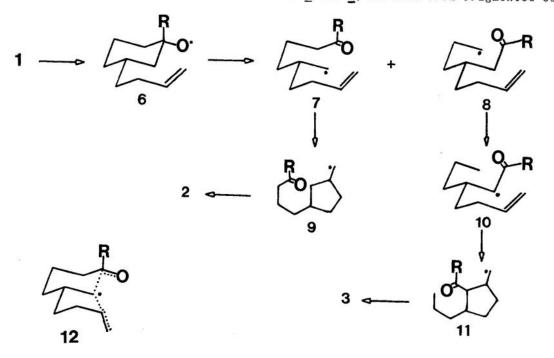
It was found that this free radical fragmentation-cyclization reaction involving a ring reconstruction can be achieved by ferrous ion induced decomposition of 1-alky1- or 1-ary1-3-(3-alkeny1)-cyclohexy1 hydroperoxides $\underline{1a-c}^{10}$. Two saturated ketones having a cyclopentane ring $\underline{2a-c}$ and $\underline{3a-c}$ were obtained as a chief reaction products in 28-38% and 7-30% yields, respectively, in addition to the unsaturated ketones $\underline{4a-c}$ (8-30%) and starting alcohols $\underline{5a-c}$ (7-20% yields) (Scheme 2.). In a typical experiment 2.48 g (13.5 mmole) of alkyl hydroperoxide \underline{la}^{11} was treated with 3.75 g of powdered crystaline ferrous sulfate in acetic acid as a solvent in inert atmosphere. During 3 hrs hydroperoxide was completly reduced. The reaction mixture was worked up as it was described¹² and products were separated and purified by <u>glc</u> and characterized by ir, nmr and mass spectra¹³.



Scheme 2.

For example, by decomposition of 1-methyl-3-(3-butenyl)cyclohexyl hydroperoxide <u>la</u> by ferrous ion 5-(3-methylcyclopentyl)-pentan-2-one <u>2a</u> (38%) and methyl (2-methyl 5-<u>n</u>-propyl)-cyclopentyl ketone <u>3a</u> (30%) were obtained as a products of ring reconstruction reactions.

Tertiary 1-alkylcycloalkyloxy radicals, regardless of its precursors, undergo to the β -cleavage reaction and 5-carbonylpentyl radical was formed as an intermediate $^{4-6}$. By possessing an alkenyl group in position 3 of cyclohexane ring the alkoxy radical <u>6</u> is not symmetrical and two possibilities for $C_{\alpha} - C_{\beta}$ bond scission exist, thus two fragmented carbon centered radicals <u>7</u> and <u>8</u> were generated as intermediates (Scheme 3.). Ratio of ketones <u>2</u> and <u>3</u>, derived from fragmented carbon



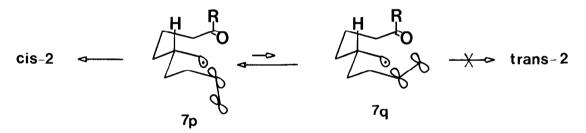
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radicals $\underline{7}$ and $\underline{6}$, respectively, indicates that a small long range participation of olefinic double bond on the course of β -cleavage reaction exists. Participation of olefinic bond on the course of β -scission reaction can be explained by involving transition state $\underline{12}$. Although, a litle lower yields of ketone $\underline{3}$ in respect to ketone 2 may be due to side reactions of radicals 8, 10 and 11.

The further reaction of these two primary, fragmented, carbon radicals $\frac{7}{1}$ and $\frac{8}{2}$ are considerably different. By possessing an olefinic bond in position 5 radical $\frac{7}{1}$ undergoes almost exclusively to the intramolecular $\frac{5-\text{exo-trigonal}}{1-\text{poss}}$ cyclization, thus producing a cyclopentylmethyl radical $\frac{9}{2}$. Termination process for radical $\frac{9}{2}$ is hydrogen abstraction and ketone $\frac{2}{2}$ was formed as a final product of radical recombination reactions. Products of $\frac{5-\text{endo-trigonal}}{1-\text{poss}}$ cyclization, coupling dicarbonyl compounds and hydrogen abstraction products, which could be derived from fragmented radical 7, were not observed.

However, fragmented primary carbon radical $\underline{8}$ undergoes to the 1,5-bydrogen abstraction from the methylene group adjacent to carbonyl group and a secondary radical <u>10</u> was generated (Scheme 3.). This radical rearrangement is energetically favourable for about 6 kcal/mole¹⁴. The fate of **c**-carbonyl radical <u>10</u> with an olefinic bond in position 5, as it was expected, it undergoes to the intramolecular cyclization reaction and a new cyclopentylmethyl radical <u>11</u> arising which affords a ketone <u>3</u> as a final product¹⁵.

Saturated ketone 2 was obtained as a pure <u>cis</u>-isomer, what is in agreement with greater stability of <u>cis</u>-1,3-dialkylcyclopentane derivatives¹⁶, as well as more favourable conformation of butenyl group in carbon radical <u>7p</u> (Scheme 4.), with quasi-endo-orientation of olefinic bond leading to <u>cis</u>-isomer <u>2</u>. While the



Scheme 4.

conformation $\underline{7q}$ which is requested for \underline{trans} -isomer formation with quasi- \underline{exo} -orientation of alkenyl group is less favourable.

Investigation of other type of alkoxy radical precursors and other type of rings reconstructions are in progress.

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