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SUPPLEMENTARY MATERIAL TO Study of pyrolysis of high density polyethylene in the open system and estimation of its capability for co-pyrolysis with lignite

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ADDITIONAL INTRODUCTORY CONSIDERATIONS

Seo *et al.* $(2003)^{1}$ reported similar or even higher yield of the liquid product obtained by the thermal cracking than by the thermo-catalytic cracking (ZSM-5 and Zeolite Y) of HDPE at 450 °C. However, the composition of liquid products was quite different. The liquid product obtained by the non-catalytic cracking contained 56 % of light hydrocarbons (C_6-C_{12}), whereas the liquid products from the thermo-catalytic cracking comprised more than 90 % of light hydrocarbons. Miskolczi *et al.* $(2004)^2$ studied the thermal and thermo-catalytic cracking behaviour of a commercial HDPE waste in the temperature range of 400-450 °C. The fluid catalytic cracking catalyst (FCC), HZSM-5 and a clinoptilolite, containing rhyolitic tuff, were used as catalysts. The yield of gases increased in the following order: non-thermo-catalytic cracking < clinoptilolite < FCC < HZSM-5 catalysed cracking, while the yields of liquid products increased in the sequence of non-thermo-catalytic cracking < clinoptilolite < HZSM-5 < FCC catalysed cracking. Barbarias et al. (2015)³ used the spent FCC catalysts regenerated with air at 550 °C for 1 h and then agglomerated by the wet extrusion with bentonite (50 wt. %) in the two-step system (flash pyrolysis at 500 °C and catalytic cracking in the temperature range from 450 to 550 °C). It was observed that an increase in the catalytic cracking temperature from 450 to 550 °C resulted in the

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significant increase of the yield of light olefins (from 12 to 28 wt. %) and non-aromatic C_5-C_{11} compounds (from 34 to 52 wt. %). Almustapha and Andresen (2011)⁴ used two sulphate modified zirconium catalysts (CAT) in the catalytic degradation of HDPE. Heating of HDPE with catalysts to 400 °C at heating rate of approximately 20 °C min⁻¹ resulted in the extremely high conversion of HDPE (about 99.8 %) into gaseous and liquid products. García *et al.* (2005)⁵ demonstrated the advantage of utilisation of hybrid zeolitic-mesoporous material (ZSM-5/MCM-41) for the catalytic degradation of HDPE. By the heating of HDPE of about 70 % was achieved. The light hydrocarbons (C_2-C_6) rich in olefins were the main degradation products, while the production of heavy fractions was negligible.

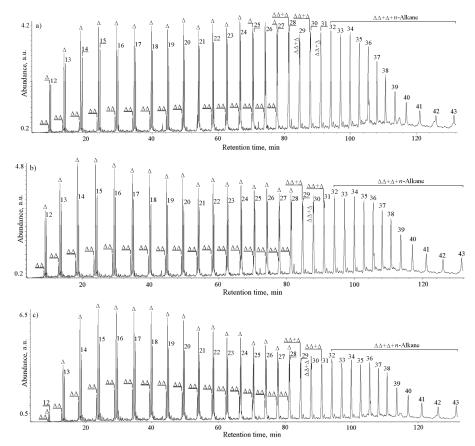


Fig. S-1. TIC (Total Ion Current) of liquid pyrolysates obtained at 400 °C (a), 450 °C (b) and 500 °C (c). *n*-Alkanes are labelled according to their carbon number; $\Delta - 1$ -*n*-alkenes with same number of carbon atoms as *n*-alkanes; $\Delta\Delta$ – Terminal dienes with same number of carbon atoms as *n*-alkanes.

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