

Scientific paper

Prediction of *in vivo* Bioavailability by *in vitro* Characterization of Ethylenediamine Dipropanoic Acid Derivatives with Cytotoxic Activity

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Abstract

O,O'-diethyl-(*S,S*)-ethylenediamine-*N,N'*-di-2-(3-cyclohexyl)propanoate (DE-EDCP) is novel substance with cytotoxic activity in human leukemic cells. The aim of this study has been to predict *in vivo* bioavailability of the DE-EDCP and its potential metabolite (*S,S*)-ethylenediamine-*N,N'*-di-2-(3-cyclohexyl)propanoic acid (EDCP) by *in vitro* characterization which includes determination of lipophilicity and passive membrane permeability. There has also been evaluated inter-laboratory reproducibility of the bio-analytical method which was previously developed and validated for non-clinical study of the DE-EDCP and EDCP.

Distribution coefficient *n*-octanol/water was 1.68 and 0.03, and apparent permeability coefficient was 4×10^{-4} cm/s and 20×10^{-4} cm/s, for the DE-EDCP and EDCP, respectively.

Observed results have shown that the DE-EDCP is more lipophilic with better membrane retention, but the EDCP has better pass through the membrane. Also, there has been demonstrated a reproducibility and robustness of the proposed bio-analytical method.

Keywords: Transfer of the UHPLC-MS/MS, cross validation, (*S,S*)-ethylenediamine-*N,N'*-di-2-(3-cyclohexyl) propanoic acid esters, membrane permeability, lipophilicity

1. Introduction

Cytotoxic activity of the novel ester derivatives of the (*S,S*)-1,2-ethanediamine-*N,N'*-di-2-(3-cyclohexyl)propanoic acid has been previously proven by the *in vitro* studies on various leukemic cell lines. It was demonstrated that methyl, ethyl, and *n*-propyl esters are toxic to HL-60, REH, MOLT-4, KG-1, JVM-2, and K-562 leukemic cell lines, while the non-esterified compound and the *n*-butyl ester are devoid of cytotoxic action. The *O,O'*-diethyl-(*S,S*)-ethylenediamine-*N,N'*-di-2-(3-cyclohexyl)propanoate dihydrochloride (DE-EDCP), has showed the highest cytotoxic activity on leukemic cell line HL-60 (IC₅₀ in the range of 11 μ M – 45 μ M). Demonstrated data show that the toxicity

is mediated by the caspase-independent apoptosis associated with oxidative stress, mitochondrial dysfunction, and AIF translocation.¹ DE-EDCP has been chosen for further characterization since it had exerted the strongest cytotoxic activity in HL-60 cell line.

In vitro characterization of new pharmaceutical substances includes determination of a lipophilicity and passive membrane permeability. These physicochemical properties of pharmaceutical substances are providing significant information for prediction of the *in vivo* bioavailability by exploring absorption and distribution behaviour of the substance. Lipophilicity is one of many factors involved in biological activity of a drug, and it is often one of

the most influential.² Lipophilicity is usually expressed by the *n*-octanol/water partition coefficient ($\log P$) for neutral molecules and the distribution coefficient ($\log D$) for ionized molecules.³ Method for a determination of the $\log D$ is based on determination of the *n*-octanol/water partition coefficient.² This procedure requires the measurement of the compound concentration in *n*-octanol and water phases after equilibration of both phases according to Eq. (1). Thus, the Eq. (1) can be written as:⁴

$$\log D = \log (c_{\text{octanol}}/c_{\text{water}}) \quad (1)$$

where c_{octanol} and c_{water} are the concentrations of a substance in *n*-octanol and aqueous phase of the partition, respectively.

Method for the *in vitro* prediction of passive membrane permeability that can be used, is the parallel artificial membrane permeability assay (PAMPA). This method is used extensively for the early drug candidate evaluation. PAMPA was first introduced by Kansy et al.^{5–8}

This method has been shown useful in assessing trans-membrane, non-energy dependent, and diffusion of drugs in such a way that a reasonable predictability with *in vivo* (passive) absorption is possible.

The artificial membrane permeability may be expressed either as a percent of transport (%*T*) or as an apparent permeability coefficient P_{app} .

$$\%T = 100 \cdot (A_{\text{R}} \cdot V_{\text{R}}) / (A_{\text{D}_0} \cdot V_{\text{D}}) \quad (2)$$

Where A_{D_0} and A_{R} are the peak areas of the initial donor solution and the post-incubation receiving solution (from the acceptor wells), V_{R} and V_{D} are the volumes of the receiving and donor solutions.

The %*T* is related to P_{app} based on the following equation:

$$P_{\text{app}} = (V_{\text{D}} \cdot V_{\text{R}}) / ((V_{\text{D}} + V_{\text{R}}) \cdot S \cdot t) \cdot \ln[(100 \cdot V_{\text{D}}) / (100 \cdot V_{\text{D}} - \%T(V_{\text{D}} + V_{\text{R}}))] \quad (3)$$

Where *S* is the surface area of the artificial membrane and *t* is the incubation time.⁵

Generally, compounds that have a $P_{\text{app}} < 10 \times 10^{-6}$ cm/s are classified as low permeability and ones with a $P_{\text{app}} > 10 \times 10^{-6}$ cm/s are classified as high permeability.

In vivo characterization of new pharmaceutical substances includes non-clinical study on animal model. For non-clinical study of cytotoxic activity of the DE-EDCP and its potential metabolite (S,S)-ethylenediamine-*N,N'*-di-2-(3-cyclohexyl)propanoic acid dihydrochloride (EDCP), there has been previously developed and validated the ultra-high performance liquid chromatography tandem mass spectrometry (UHPLC-MS/MS) bio-analytical method.⁹ Bioanalytical methods should be robust.¹⁰ Evaluation of reproducibility (transferability) of bio-analytical method is becoming increasingly important^{11–13} since

the bio-analytical methods are often used in different laboratories during non-clinical and clinical studies. The transfer process requires the procedure to be physically transferred from a laboratory which masters the technique (called sender or originator) to another site (called receiver or recipient).¹⁴ In the context of bio-analysis, method transfer is covered by the Food and Drug Administration (FDA) and European Medicines Agency (EMA) guidance documents on bio-analytical method validation.^{11,13,15} Although the need for method transfer is recognized by both authorities, little is said about the process itself. Several approaches have been described for the method transfers evaluation.^{12,16–24}

New UHPLC-MS/MS bio-analytical method which has been developed and validated for the *in vivo* characterization (non-clinical study) of the DE-EDCP and EDCP⁹ is supposed to be used in different laboratories during the mentioned studies. However, *in vitro* characterization (lipophilicity and membrane permeability) of the DE-EDCP and EDCP has not been investigated until now.

The aim of this study is to predict the *in vivo* bioavailability of the DE-EDCP and its potential metabolite EDCP by the *in vitro* characterization – determination of lipophilicity and passive membrane permeability. In this study, there has also been evaluated the inter-laboratory reproducibility of the previously mentioned bio-analytical method.

2. Experimental

2.1. Chemicals

O,O'-diethyl-(S,S)-ethylenediamine-*N,N'*-di-2-(3-cyclohexyl)propanoate (DE-EDCP), (S,S)-ethylenediamine-*N,N'*-di-2-(3-cyclohexyl)propanoic acid dihydrochloride (EDCP) and the internal standard (S,S)-*O,O'*-dibutyl-1,3-propanediamine-*N,N'*-di-2-(3-cyclohexyl)propanoate dihydrochloride (DB-PDCP), were provided by the Faculty of Chemistry, University of Belgrade, Serbia (Table 1).

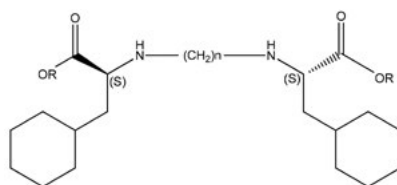
Acetonitrile, methanol, ethyl acetate, diethyl ether, triethanolamine, chloroform and trifluoroacetic acid (HPLC grade), ammonium acetate (CH₃COONH₄), sodium fluoride and KH₂PO₄ (ACS grade) from Fluka (Sigma-Aldrich Co.), *n*-octanol from Fluka AG (Buchs SG, Switzerland) and deionized water (TKA GenPure Ultrapure, Germany), were used. Mouse serum was purchased from Sigma Aldrich (Saint Louis, USA).

2.2. Solutions

Preparation of all standard solutions is described in a previously published study.⁹ Sample preparation is also described previously.⁹ All these solutions were prepared by two different analysts in sending and receiving laboratories.

2.3. Equipment

Solids were weighted by using a 5-digit Mettler analytical balance (Mettler-Toledo International Inc, USA),

Table 1: Structure of EDCP, DE-EDCP, and the internal standard DB-PDCP

Name of compound	Empirical formula	Abbreviation	MW (g/mol)	R	n
(S,S)-ethylenediamine- <i>N,N'</i> -di-2-(3-cyclohexyl)propanoic acid dihydrochloride	C ₂₀ H ₃₈ O ₄ N ₂ Cl ₂	EDCP	441.43	H	2
<i>O,O'</i> -diethyl-(S,S)-ethylenediamine- <i>N,N'</i> -di-2-(3-cyclohexyl)propanoate dihydrochloride	C ₂₄ H ₄₆ O ₄ N ₂ Cl ₂	DE-EDCP	497.54	C ₂ H ₅	2
(S,S)- <i>O,O'</i> -dibutyl-1,3-propanediamine- <i>N,N'</i> -di-2-(3-cyclohexyl)propanoate dihydrochloride	C ₂₉ H ₅₆ O ₄ N ₂ Cl ₂	DB-PDCP	567.67	C ₄ H ₉	3

and less sensitive weighting was performed on the Adventurer Pro analytical balance (OHAUS, USA). Sample preparation was done by using the Eppendorf 5417R micro-centrifuge (Eppendorf, Germany).

2. 4. Sending Laboratory

Development and validation of the method was done in the sending laboratory on the Thermo ACCELA (Thermo Scientific, Waltham, Massachusetts, USA) UHPLC system, coupled to a triple quad Mass Spectrometer Thermo TSQ Quantum Access Max (Thermo Scientific, Waltham, Massachusetts, USA), with a heated electro-spray ionization (HESI) interface.²⁴ A reverse-phase Thermo Scientific Hypersil GOLD aQ column (100 × 2.1 mm 1.9 μm ThermoScientific, and guard cartridge (Thermo Scientific Hypersil GOLD aQ, 10 mm l × 4 mm ID), were used in both laboratories.

2. 5. Receiving Laboratory

In the receiving laboratory, method transfer and validation were carried out on the Agilent 1290 UHPLC system equipped with the Agilent 6420 triple-quad mass detector (Agilent Technologies, Santa Clara, USA) with the electro-spray ionization (ESI) interface.

DE-EDCP, EDCP and IS were eluted by using a mobile phase as previously described.⁹ Quantitation was achieved by the MS–MS detection in the positive ionization mode for the DE-EDCP, EDCP and IS. The MS operating conditions were optimized as following: the capillary voltage was 4500 V, the gas temperature was set to 340 °C and gas flow was 10 l/min and the nebulizer pressure was 35 psi. Nitrogen was used as a collision gas. Fragmented voltage was set to 135 V. Ions detection was performed in the multiple reaction monitoring (MRM) by using the following transitions of *m/z* 425.2 → 197.8 and 226.1 for DE-EDCP, *m/z* 369.3 → 152.1 and 198.1 for EDCP and *m/z*

495.3 → 166.0 and 268.1 for DB-PDCP (IS), respectively, with a scan time of 0.1 s per transition.

Mass Hunter Optimizer software version 6.00 (Agilent Technologies, Santa Clara, USA) was used for automatic optimization of the acquisition parameters. Data Acquisition was performed by using the Mass Hunter Data Acquisition software version 6.00 (Agilent Technologies, Santa Clara, USA).

Qualitative and quantitative data analyses were done by using the Mass Hunter Qualitative software version 6.00 and Mass Hunter Quantitative software version 6.00 (Agilent Technologies, Santa Clara, USA), respectively.

Method transfer was done from the sending laboratory to the receiving laboratory.

The cross-validation samples were analyzed through a fully validated bio-analytical method at the receiving laboratory along with the calibration of the standards and QC samples for the validity of each analytical run. The following validation parameters were evaluated in the receiving laboratory for the UHPLC-MS/MS system: selectivity, linearity, limit of quantification (LLOQ), recovery, accuracy, precision and matrix effects.

Also, cross-validation samples were analyzed by the Passing and Bablok regression analysis. Passing and Bablok regression analysis is a statistical procedure which allows the valuable estimation of the analytical methods agreement and possible systematic bias between them. Results are presented with a scatter diagram and a regression line, as well as a regression equation where an intercept represents a constant and slope proportional measurement error. Confidence intervals of 95% of the intercept and slope, give the explanation whether their value differ from the value zero (intercept) and value one (slope) only by chance, allowing a conclusion of the method agreement and a correction action, if necessary.²⁵

During this study, there have been tested the selectivity, linearity, limit of quantification (LLOQ), recovery (%), matrix effects, accuracy and precision, as described previously.⁹

Cross-validation showed to be successful in terms of the results' traceability between the two instruments (slope and intercept with confidential interval values) and the results of validation parameters (selectivity, linearity, limit of quantification (LLOQ), recovery, accuracy, precision and matrix effects).

2. 6. Determination of the *n*-octanol/water Distribution Coefficient (Log*D*)

Log*D* values were determined by using a shake-flask method. In the shake-flask experiment, 5 mg of each substance (DE-EDCP and EDCP) was first mixed with 50 ml of aqueous buffer (pH 7.4). Then, 10 ml of this solution was mixed with 10 ml of the *n*-octanol (water saturated). The sample vial was placed on the shaker and been shaken for 12 h at 250 rpm. After equilibration, it was left to stand for 2 h to phases well separated.

The separated aqueous phase is being centrifuged, the residual drops of the *n*-octanol to be eliminated. The aqueous phase was sampled and assayed by the transferred and cross-validated UHPLC-MS/MS which had been previously validated for the determination of the investigated substances in the aqueous buffer (pH 7.4) in order to determine the log*D* value. The concentration of the investigated substances in the *n*-octanol phase was obtained as a difference in the concentrations in the aqueous buffer, prior to mixing with the *n*-octanol and after mixing with the *n*-octanol.

2. 7. Prediction of Membrane Permeability (PAMPA test)

The *in vitro* method for the prediction of membrane permeability which was used in these studies, was carried out in a 96-well format. 96-well micro-titer plates (hydrophobic PVDF MultiScreen IPFilter Plate 0.45 μm , from Milipore (Bedford, MA, USA)), were assembled into such a "sandwich" that each composite well was separated by a 125 μm micro filter disc. Filter material in each well of the filtration plate was wetted with 5 μl of the artificial membrane solution, which consisted of 1 % egg lecithin in the *n*-dodecane. Subsequently, the filter plate was placed on the bottom micro-titer plate containing the following donor solution: 300 μl of the compound in the concentration of 0.1 mg/ml dissolved in the buffer KH_2PO_4 0.2M, pH = 7.4. The top acceptor wells of the sandwich, were hydrated with the 300 μl of the buffer KH_2PO_4 0.2M, pH 7.4. To prevent loss by evaporation, the system was first covered with a paraffinic film. The surface area of the artificial membrane was $S = 0.28 \text{ cm}^2$ and the period of incubation was $t = 7200 \text{ s}$ (2 h).

After incubation, the amount of the DE-EDCP and EDCP in the donor and acceptor wells, was determined by the UHPLC-MS/MS method which had been previously transferred and cross-validated.

2. 8. Software

For the determination of lipophilicity (Log*D*_{7,4}) of the investigated substances by the *in silico* model, there was used the MarvinSketch 4.1.13 (ChemAxon, Budapest, Hungary).

3. Results and Discussion

3. 1. UHPLC-MS/MS Method: Transfer and Cross-validation

Transfer of the ultra-high performance liquid chromatography-electro-spray tandem mass spectrometry (UHPLC-ESI-MS/MS) method for non-clinical studies and the *in vitro* characterization of recently synthesized substances with the cytotoxic activity, DE-EDCP and its potential metabolite EDCP, in biological material, was carried out. The reproducibility and transferability of this bio-analytical method in the mouse serum was evaluated by the validation and cross-validation of the method through using two different UHPLC-MS/MS systems. The parallel displayed values of the observed validation parameters are given in the Table 2.

The method was proven to be highly selective for the analytes, since no interfering peaks from the endogenous compounds were observed at the retention times for the DE-EDCP and EDCP in any of the six independent blank serum extracts evaluated.

Also, cross-validation samples analyzed by the Passing and Bablok regression analysis, showed to be successful in terms of the results' traceability between the two instruments (for DE-EDCP: slope = 0.9821 with lower 95 %-CL = 0.7737 and upper 95 %-CL = 1.1685, and intercept = 0.0547 with lower 95 %-CL = -5.1662 and upper 95 %-CL = 3.4285; for EDCP: slope = 1.0187 with lower 95 %-CL = 0.9522 and upper 95 %-CL = 1.1505, and intercept = -0.0488 with lower 95 %-CL = -0.5347 and upper 95 %-CL = 0.2187).

Overall results of the cross-validation were satisfactory in terms of all the investigated parameters proving that the method can be successfully transferred under the aforementioned conditions. Results of the validation and cross validation demonstrate that the novel UHPLC-MS/MS method for the *in vivo* characterization (non-clinical study) of the novel DE-EDCP and EDCP substances with cytotoxic activity, is appropriately transferred and validated at the receiving laboratory.

3. 2. Validation of the UHPLC-MS/MS in the Aqueous Buffer

The observed UHPLC-MS/MS method was successfully validated for the determination of the investigated substances DE-EDCP and EDCP from an aqueous buffer (pH 7.4). Results of the validation parameters are given in the Table 3.

Table 2: Validation parameters for the DE-EDCP and EDCP on the Thermo ACCELA and the Agilent 1290 UHPLC system

Validation parameter	Sending laboratory (Thermo ACCELA)		Receiving laboratory (Agilent 1290 UHPLC system)	
	DE-EDCP	EDCP	DE-EDCP	EDCP
Linearity of calibration curves	1.3–26.7 ng/ml $y = 0.0461x + 0.0895$, $r = 0.9978$	0.33–6.67 µg/ml $y = 0.1527x + 0.0045$, $r = 0.9987$	3.3–26.7 ng/ml $y = 64.243x + 114.54$, $r = 0.9983$	0.33–6.67 µg/ml $y = 9322.3x - 103.4$, $r = 0.9989$
LLOQ	1.3 ng/ml	0.33 µg/ml	3.3 ng/ml	0.33 µg/ml
Recovery %	90.0–99.3	75.8–100.3	91.0–99.8	77.8–101.5
Matrix effect	95.5–108.2%		96.7–109.4%	
Precision (%CV)	15.99	5.58	3.22	4.00
	13.68	4.43	3.36	2.05
	2.25	5.32	3.97	3.95
	3.49	4.01	1.87	2.00
Accuracy (%RE)	3.01	6.06	17.05	-9.93
	12.61	-2.41	11.50	-14.52
	6.30	-3.20	5.20	-13.13
	1.80	-14.40	1.73	-14.20

y represents the peak area ratio of analyst to IS

Table 3: Validation parameters for the DE-EDCP and EDCP in an aqueous buffer, without a biological matrix (Thermo ACCELA)

Validation parameter	Thermo ACCELA	
	DE-EDCP	EDCP
Linearity of calibration curves	2.0 to 40.0 ng/ml $r = 0.9930$	0.5 to 10.0 µg/ml $r = 0.9997$
LLOQ	2.0 ng/ml	0.50 µg/ml
Precision (%CV)	3.29 to 18.07%	6.09 to 15.50%
Accuracy (%RE)	0.01 to 16.00%	1.48 to 13.50%

3. 3. Determination of the Log $D_{7.4}$ *in vitro/in silico* and the PAMPA Test

Lipophilicity of the observed substances DE-EDCP and EDCP, has been tested by the traditional shake-flask method. Passive membrane permeability has been tested by the PAMPA test. Experimental results that have been obtained in this study are shown in the Table 4. Also, in the Table 4, there can be seen *in silico* results of the lipophilicity of the DE-EDCP and EDCP.

Table 4: Lipophilicity and passive membrane permeability data for the investigated substances DE-EDCP and EDCP

Compound	Log $D_{7.4}$ *	Log $D_{7.4}$	P_{app} (cm/s)	%T	% Membrane retention
EDCP	-1.8	0.03	$20 \cdot 10^{-4}$	7.94	6.90
DE-EDCP	4.04	1.68	$4 \cdot 10^{-4}$	1.76	97.46

* by MarvinSketch 4.1.13

The results represent obvious difference in the lipophilicity between the DE-EDCP and EDCP. Lipophilicity data obtained through the shake flask method, and membrane permeability data acquired by the PAMPA

test, are correlated with the results gained in the previous *in vitro* activity studies on various leukemic cell lines of the investigated compounds. Compound DE-EDCP with a significant cytotoxic activity,¹ has greater lipophilicity which allows more retention in the cell membrane. This characteristic of the aforementioned compound is particularly important for its activity. On the other hand, the suspected metabolite EDCP is more hydrophilic and passes through the membrane, without retention in the cell membrane.

4. Conclusions

In this study, DE-EDCP and EDCP bioavailability was examined through *in vitro* characterization, by determination of lipophilicity and passive membrane permeability. Observed results are showing that the DE-EDCP is more lipophilic than the EDCP, with better membrane retention.

Additionally, this study proved good reproducibility (transferability) and robustness of the bioanalytical method for *in vivo* characterization of the investigated substances.

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Povzetek

(S,S)-O, O-dietil-1,2-etandiamin-*N,N'*-di-2-(3-cikloheksil)propanoat (DE-EDCP) je nova snov s citotoksično aktivnostjo v človeških levkemičnih celicah. Cilj te študije je bil napovedovanje in vivo biološke uporabnosti DE-EDCP in njegovega potencialnega metabolita (S,S)-1,2-etandiamina-*N,N'*-di-2-(3-cikloheksil) propanojske kisline (EDCP) z in vitro karakterizacijo, ki vključuje določanje lipofilnosti in pasivne membranske prepustnosti. Ocenjena je bila medlaboratorijska obnovljivost biološke analitske metode, ki je bila predhodno razvita in potrjena za neklinično študijo DE-EDCP in EDCP.

Porazdelitveni koeficient med *n*-oktanolom in vodo je bil 1,68 in 0,03, navidezni koeficient prepustnosti pa je bil 4×10^{-4} cm/s in 20×10^{-4} cm/s za DE-EDCP in EDCP.

Opaženi rezultati so pokazali, da je DE-EDCP bolj lipofilen z boljšim zadrževanjem v membrani, medtem ko EDCP bolje prehaja skozi membrano. Lahko domnevamo, da je mehanizem citotoksične aktivnosti DE-EDCP na ravni celične membrane. Dokazana je bila ponovljivost in robustnost predlagane bioanalitične metode.